## In the Claims:

A complete listing of claims in the instant application is provided below as follows:

- 1 1. (Currently amended) A system for reducing the amount of oxygen
- 2 in an oxygen-containing gas within a closed environment,
- 3 comprising:
- 4 a source of hydrogen gas;
- 5 a controllable means for mixing, in response to a control
- 6 signal, a selected amount of said hydrogen gas with a portion of
- 7 said oxygen-containing gas from said closed environment to form a
- 8 first gas mixture that includes hydrogen and oxygen;
- 9 a catalyst coupled to said controllable means for receiving
- 10 said first gas mixture, said catalyst causing a reaction between
- 11 said hydrogen and at least a portion of said oxygen in said first
- 12 gas mixture wherein a second gas mixture is formed and returned to
- 13 said closed environment, said second gas mixture (i) having a
- 14 lower percentage of oxygen than said first gas mixture, and (ii)
- 15 containing oxygen in an amount sufficient to make said second gas
- 16 mixture breathable; and
- 17 at least one oxygen sensor positioned in said closed
- 18 environment and coupled to said controllable means for generating
- 19 said control signal when oxygen levels of said oxygen-containing
- 20  $\underline{\text{gas}}$  in said closed environment reach a threshold level  $\underline{\text{defined for}}$
- 21 an ambient pressure in said closed environment.

1 2. (Original) A system as in claim 1 wherein said source contains 2 said hydrogen gas in its pure form.

- 1 3. (Original) A system as in claim 1 wherein said source of said 2 hydrogen gas comprises a metal hydride material.
- 1 4. (Original) A system as in claim 1 wherein a volume percentage 2 of said hydrogen gas in said first gas mixture is less than 3 approximately one percent.
- 15. (Original) A system as in claim 1 wherein said catalyst is a 2 precious metal.
- 1 6. (Original) A system as in claim 5 wherein said precious metal 2 is selected from the group consisting of palladium and platinum.
- 17. (Original) A system as in claim 1 wherein said reaction caused 2 by said catalyst is defined by a reaction of one-half mole of said 3 oxygen in said first gas mixture with one mole of said hydrogen in 4 said first gas mixture to produce water vapor and heat.

1 8. (Currently amended) A system for reducing the amount of oxygen 2 in an oxygen-containing gas within a closed environment, 3 comprising:

- 4 a source of hydrogen gas under pressure;
- a controllable valve having an input and an output, said input coupled to said source, said controllable valve dispensing a variable amount of said hydrogen gas from said input to said output in accordance with a control signal;
- 9 a chamber coupled to said output of said controllable valve 10 for receiving therein said variable amount of said hydrogen gas 11 so-dispensed;
- means coupled to said chamber for drawing a portion of said 13 oxygen-containing gas from said closed environment into said 14 chamber wherein said variable amount of said hydrogen gas and said 15 portion of said oxygen-containing gas combine to form a first gas 16 mixture that includes hydrogen and oxygen;
- a catalyst coupled to said chamber for receiving said first 18 gas mixture, said catalyst causing a water vapor-producing 19 reaction between said hydrogen and at least a portion of said 20 oxygen in said first gas mixture wherein a second gas mixture is 21 formed and returned to said closed environment, said second gas 22 mixture (i) having a lower percentage of oxygen than said first 23 gas mixture, and (ii) containing oxygen in an amount sufficient to 24 make said second gas mixture breathable; and

at least one oxygen sensor positioned in said closed 26 environment and coupled to said controllable valve for generating 27 said control signal when oxygen levels of said oxygen-containing 28 gas in said closed environment reach a threshold level defined for 29 an ambient pressure in said closed environment.

- 1 9. (Original) A system as in claim 8 wherein a volume percentage 2 of said hydrogen gas in said first gas mixture is less than 3 approximately one percent.
- 1 10. (Original) A system as in claim 8 wherein said catalyst is a 2 precious metal.
- 1 11. (Original) A system as in claim 10 wherein said precious 2 metal is selected from the group consisting of palladium and 3 platinum.
- 1 12. (Original) A system as in claim 8 wherein said source of said 2 hydrogen gas under pressure is maintained outside of said closed 3 environment.

1 13. (Original) A system as in claim 8 wherein said reaction 2 caused by said catalyst is defined by a reaction of one-half mole 3 of said oxygen in said first gas mixture with one mole of said 4 hydrogen in said first gas mixture.

1 14. (Currently amended) A system as in claim 8 further comprising 2 a pressure sensor for measuring <u>said</u> ambient pressure inside said 3 closed environment, said pressure sensor coupled to said at least 4 one oxygen sensor <u>for adjusting wherein</u> said threshold level <u>is</u> 5 <u>adjusted</u> in accordance with said ambient pressure <u>so-measured</u>.

1 15. (Currently amended) A method of reducing the amount of oxygen

- 2 in an oxygen-containing gas within a closed environment,
- 3 comprising the steps of:
- 4 monitoring oxygen levels in said closed environment;
- 5 generating a control signal when said oxygen levels in said
- 6 closed environment reach a threshold level defined for an ambient
- 7 pressure in said closed environment;
- 8 mixing, in response to a control signal, a selected amount of
- 9 hydrogen gas with a portion of said oxygen-containing gas from
- 10 said closed environment to form a first gas mixture that includes
- 11 hydrogen and oxygen;
- 12 exposing said first gas mixture to a catalyst capable of
- 13 causing a reaction between said hydrogen and at least a portion of
- 14 said oxygen in said first gas mixture wherein a second gas mixture
- 15 is formed, said second gas mixture (i) having a lower percentage
- 16 of oxygen than said first gas mixture, and (ii) containing oxygen
- 17 in an amount sufficient to make said second gas mixture
- 18 breathable; and
- 19 dispensing said second gas mixture into said closed
- 20 environment.
- 1 16. (Original) A method according to claim 15 wherein said
- 2 reaction produces water vapor.

1 17. (Original) A method according to claim 15 wherein a volume 2 percentage of said hydrogen gas in said first gas mixture is less 3 than approximately one percent.

- 1 18. (Original) A method according to claim 15 wherein said step 2 of mixing comprises the steps of:
- providing an open chamber in said closed environment; and
- injecting said hydrogen gas into said open chamber under pressure to draw said oxygen-containing gas into said open chamber.
- 1 19. (Currently amended) A method according to claim 15 further 2 comprising the steps of:
- 3 measuring <u>said</u> ambient pressure inside said closed 4 environment; and
- 5 adjusting said threshold level in accordance with said 6 ambient pressure <u>so-measured</u>.